

Phosphorus sequestration by oxidizing iron in groundwater fed lowland catchments

Stijn Baken

KU Leuven

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Thanks to:

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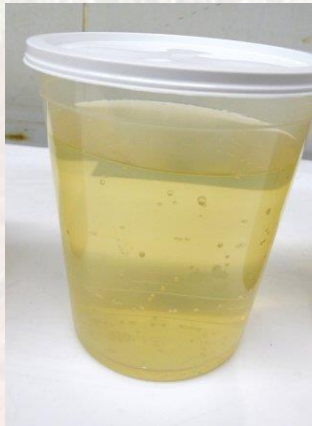
Nele Desmet – Piet Seuntjens (VITO)



Oxidation of Fe-rich groundwater → Fe oxyhydroxide particles

JUST AFTER SAMPLING

Reduced
Fe(II)
(soluble)



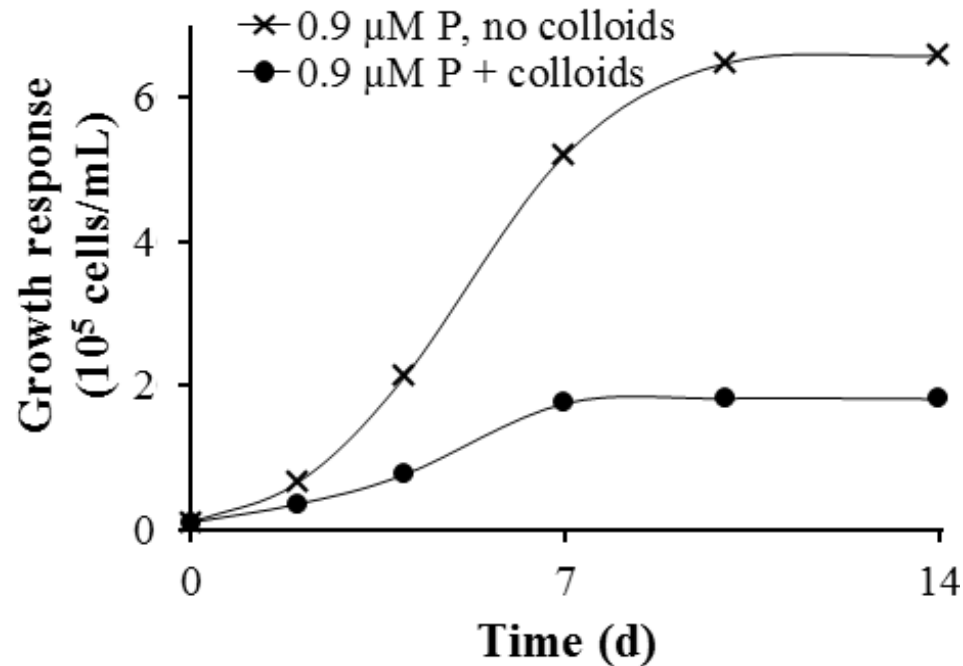
2 DAYS LATER

Oxidized
Fe(III)
(insoluble)



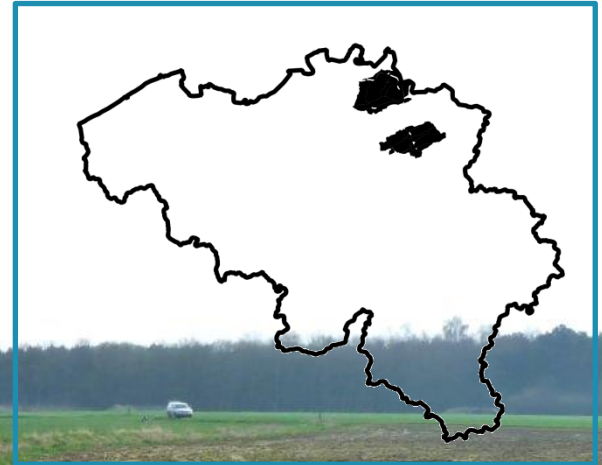
- Oxidation rate at circumneutral pH: hours...days
- In the presence of PO_4
 - First, Fe hydroxyphosphate formation until near depletion of PO_4
 - Then, continued formation of Fe oxyhydroxides

The P bound to Fe oxyhydroxides is poorly available to algae



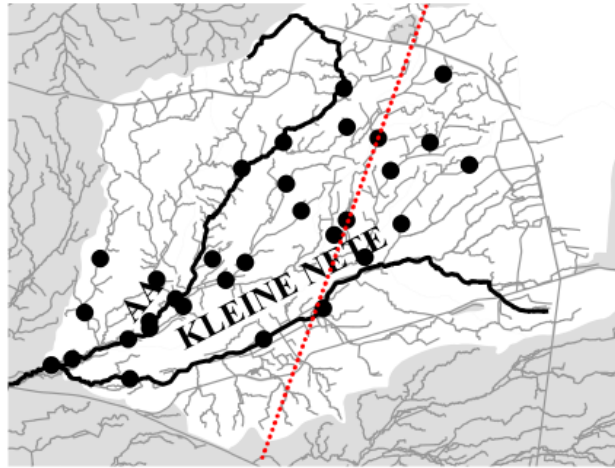
Algal growth and P uptake is reduced by adding Fe oxyhydroxide colloids.
After time, part of the P slowly becomes available.

What if Fe-rich groundwater surfaces across a large area?

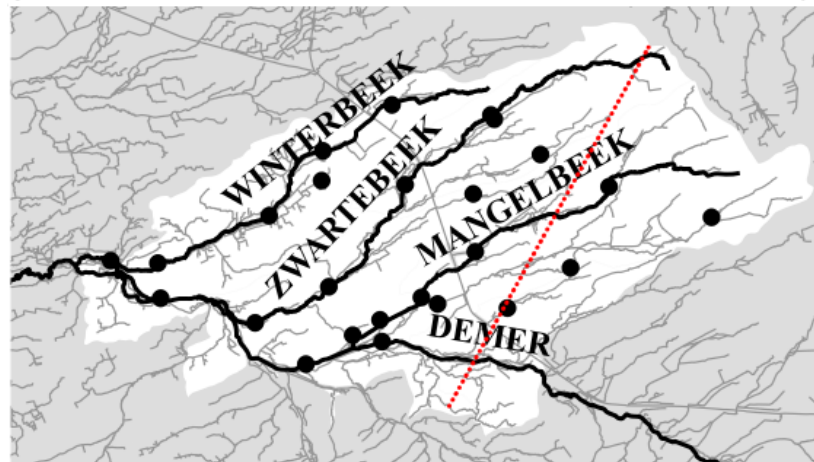


STUDY AREA

- Groundwater fed lowland catchments
- Flat topography, shallow phreatic aquifers
- Underground: permeable marine deposits, some rich in glauconite
- Average shallow groundwater composition:
20 mg Fe/L; 0.4 mg P/L
 - Fe source: glauconite weathering
 - P sources: geogenic, leaching from agricultural land

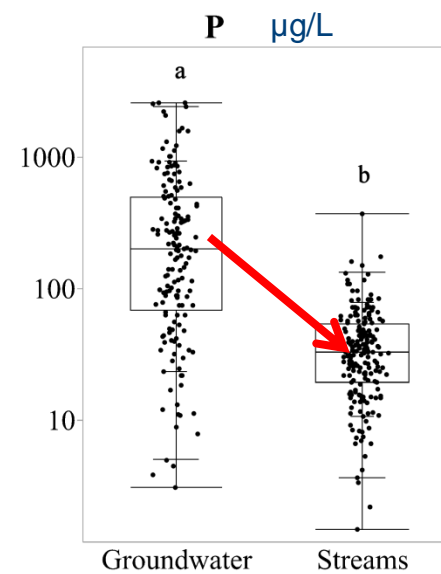
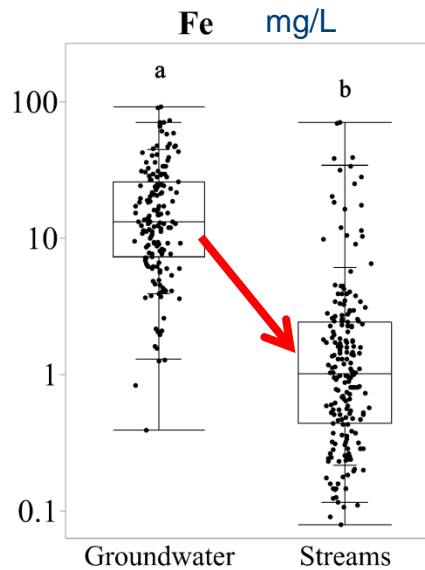
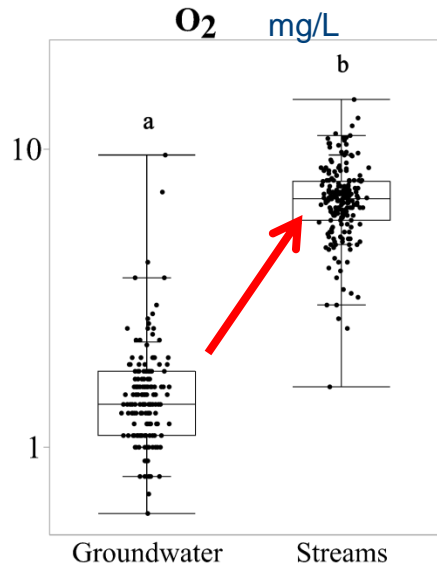
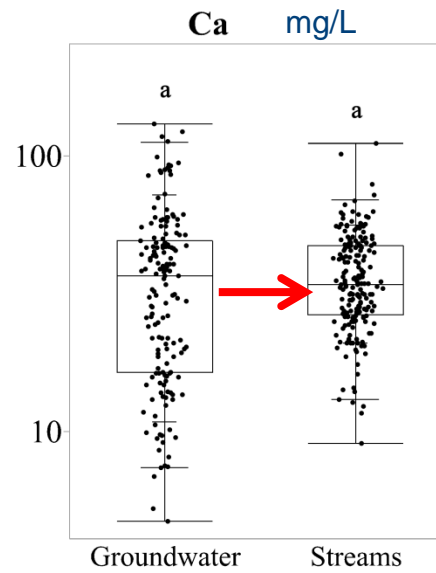
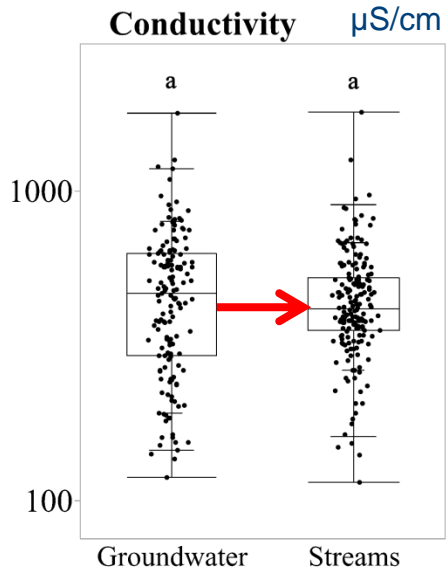


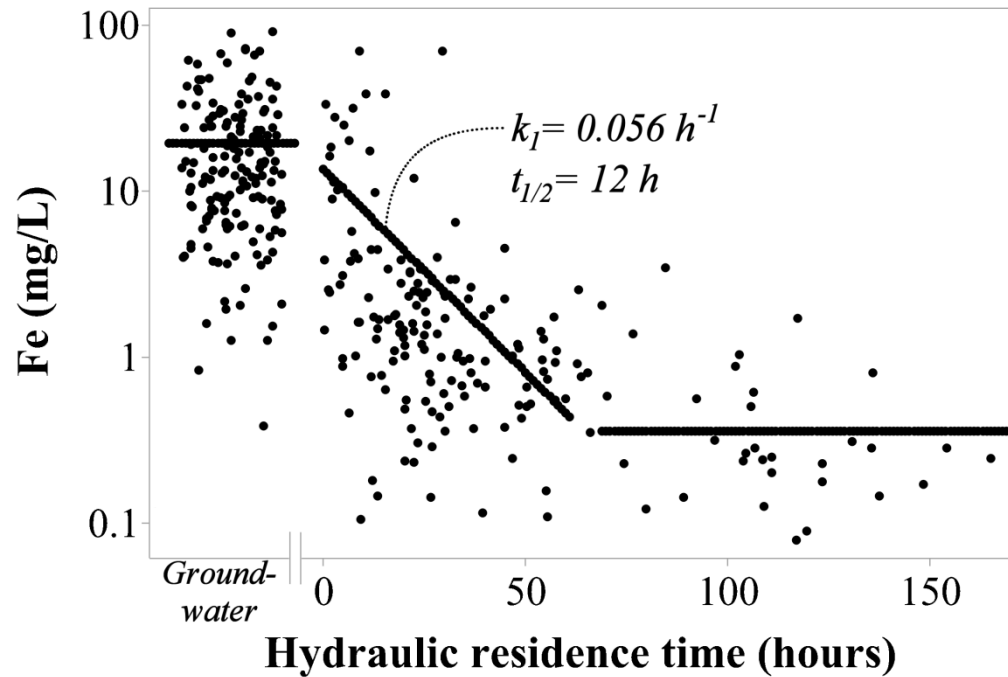
700 km²



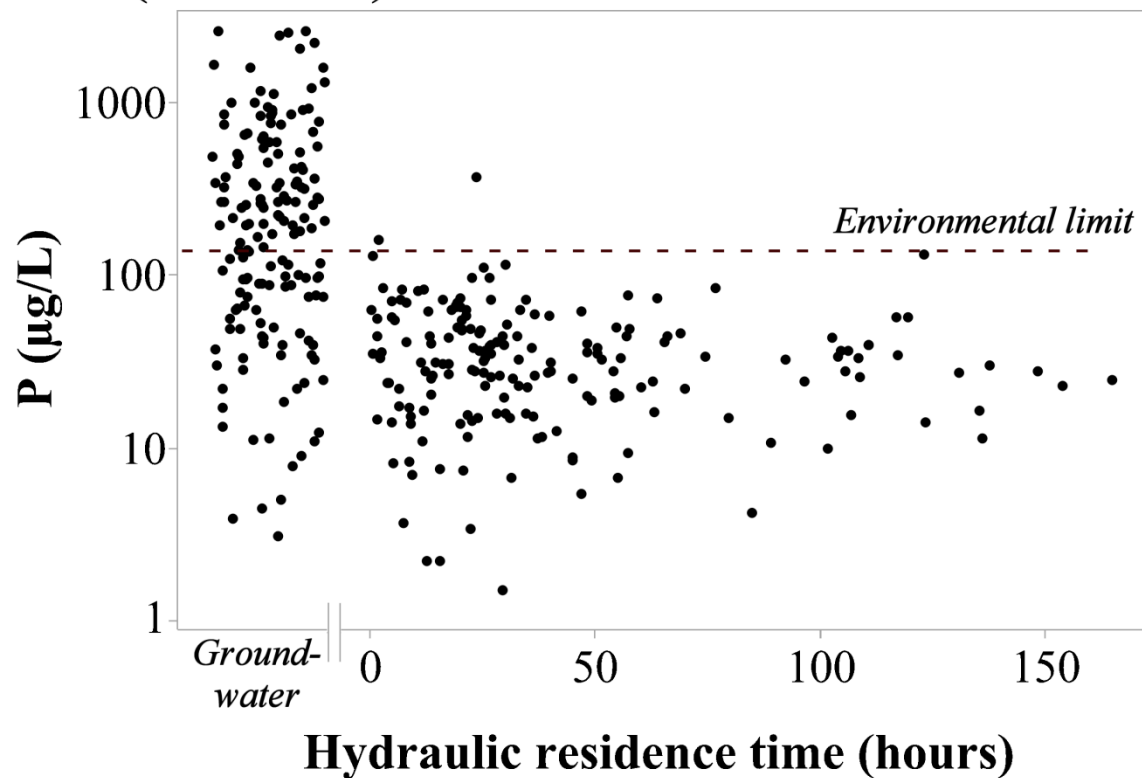
300 km²

Composition of filtered groundwater and surface water





The Fe(II) is oxidized and removed from solution ($<0.45 \mu\text{m}$) as the groundwater surfaces and flows through the catchment into increasingly larger streams

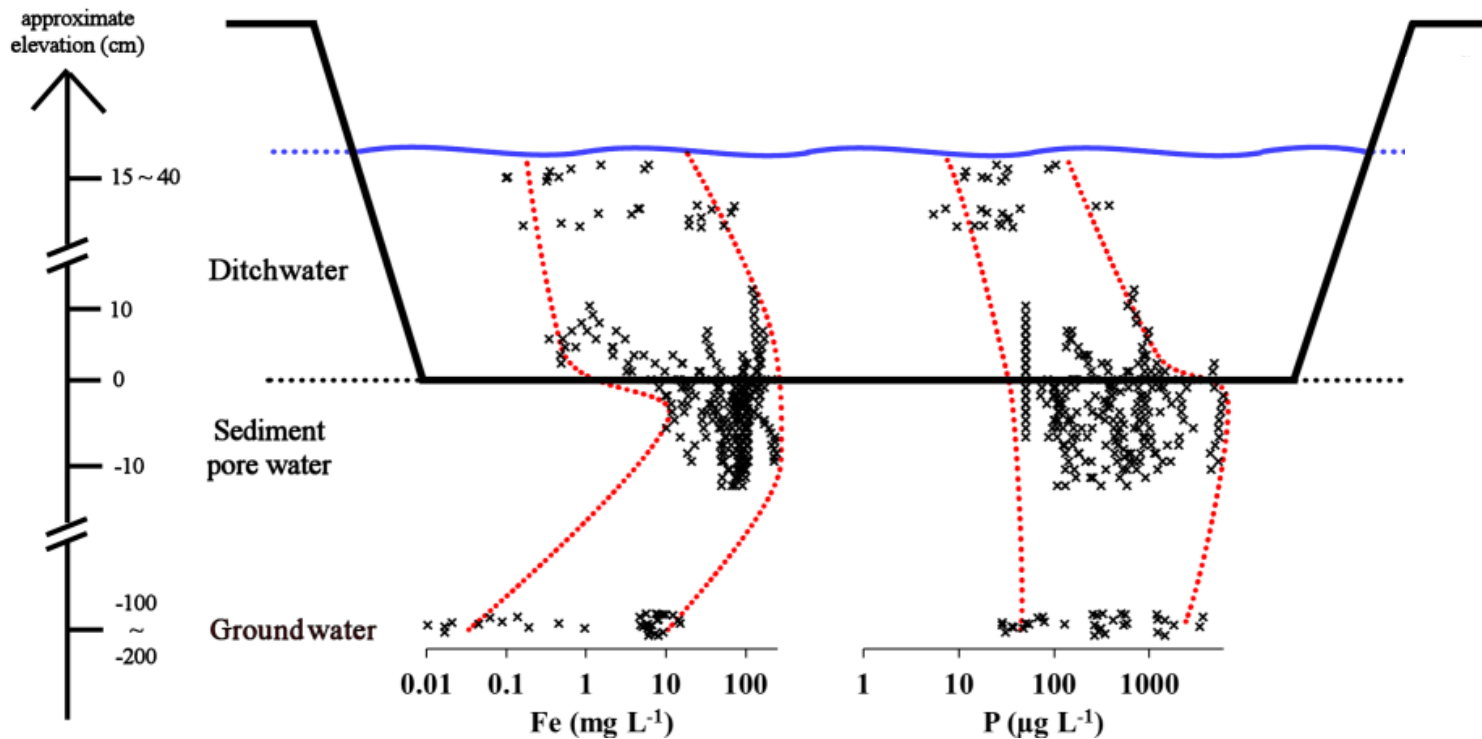


Before the water reaches the headwater streams, 90% of the P is removed from the <0.45 µm fraction

Drainage ditches with Fe-rich sediments seasonally evacuate groundwater



What happens as Fe and P bearing groundwater flows through Fe-rich sediments and into drainage ditches?



- As groundwater surfaces and flows through the catchment, the Fe is oxidized and Fe oxyhydroxide particles and colloids are produced.
- These particles strongly bind P, and thereby remove it from solution
- This study area is extreme in terms of its low P:Fe ratio
 - to what extent does this process occur in other lowlands?

